



中国科学院大连化学物理研究所  
DALIAN INSTITUTE OF CHEMICAL PHYSICS, CHINESE ACADEMY OF SCIENCES

# 大连化物所第十届青年学术报告会

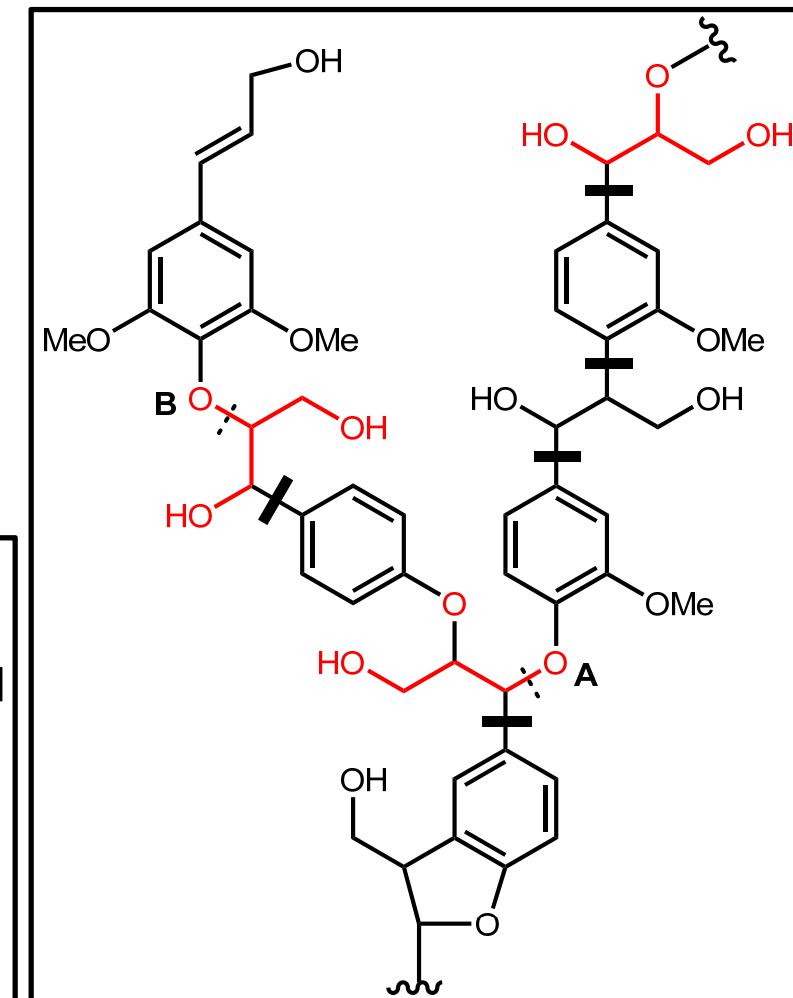
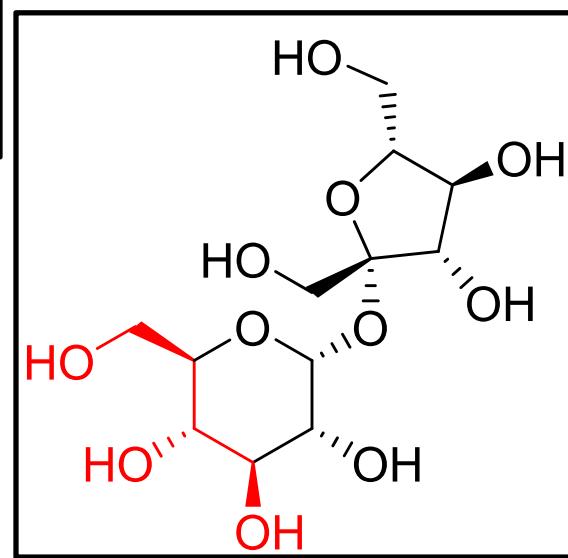
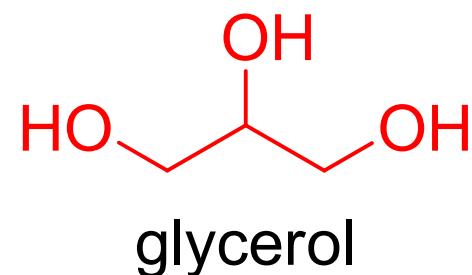
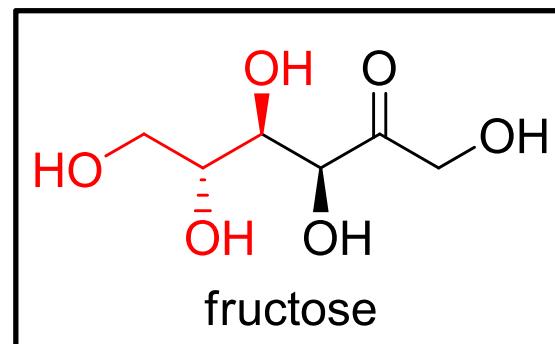
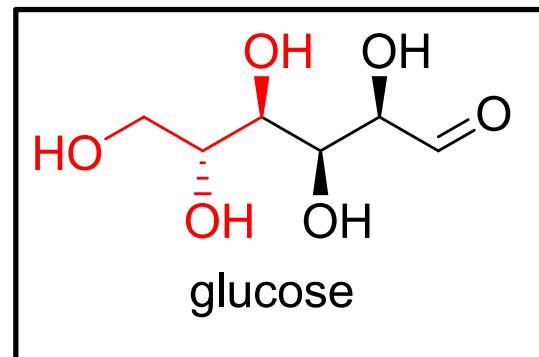
## 催化转化丙三醇制备高附加值化学品

王峰

September 29, 2010

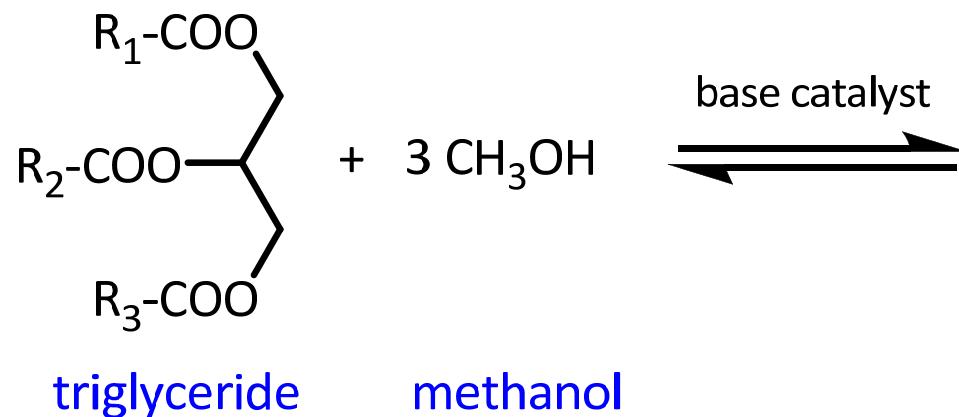


# 丙三醇是多元醇的基本结构单元

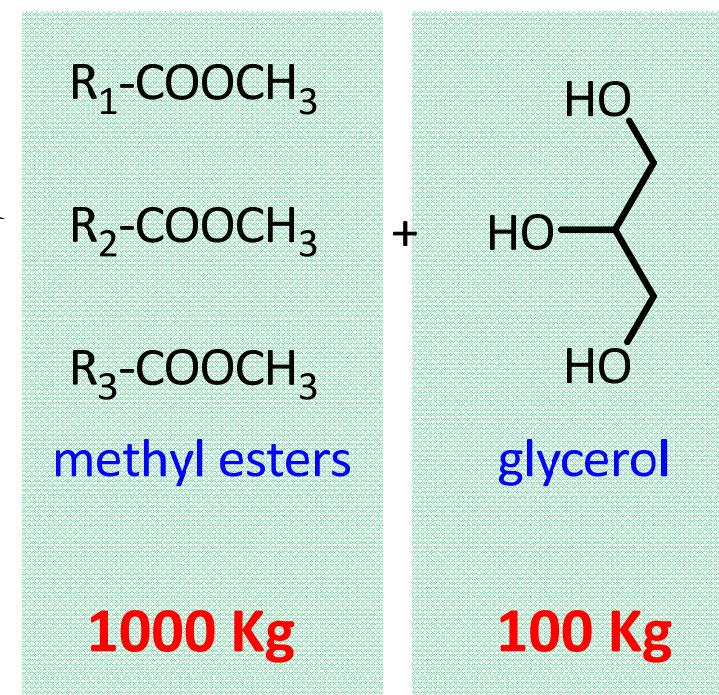


# 丙三醇是生产生物柴油的副产物

- 生物基甘油产品

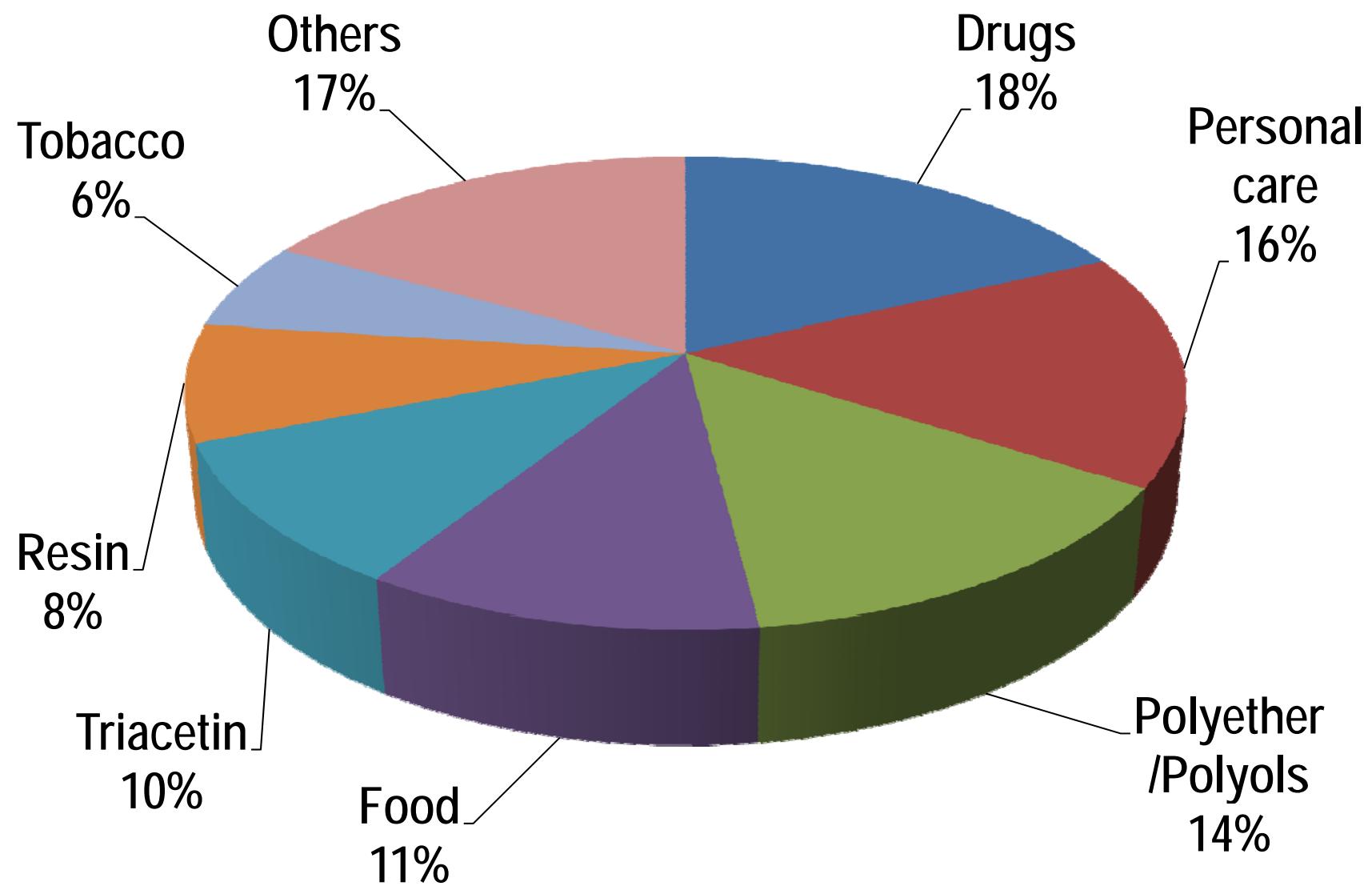


Carbon number of  $R_1$ ,  $R_2$ ,  $R_3$  is between 15 and 21

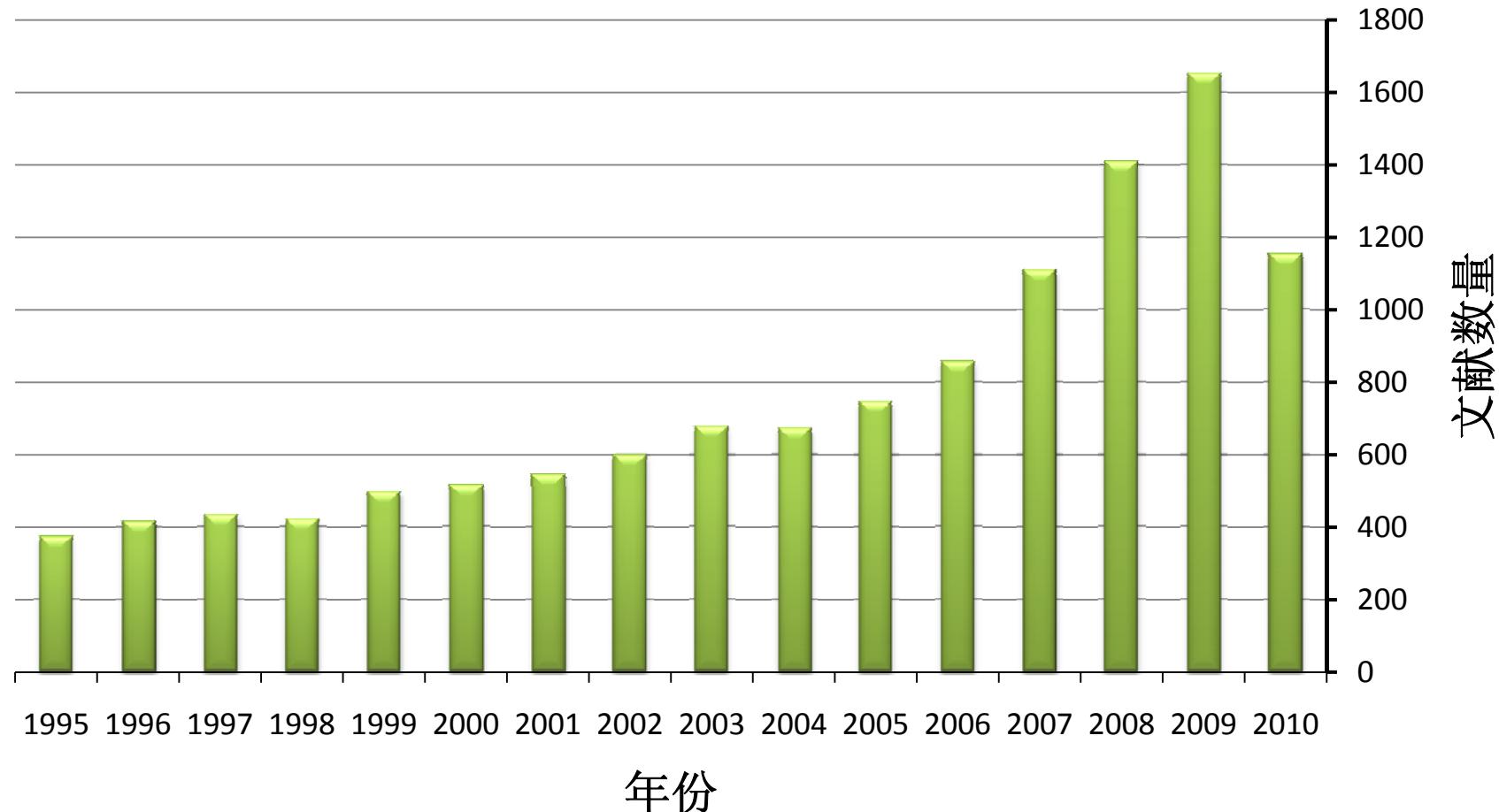


M. McCoy, *Chem. Eng. News*, 2006, **84**, 7.

# 精制丙三醇的用途

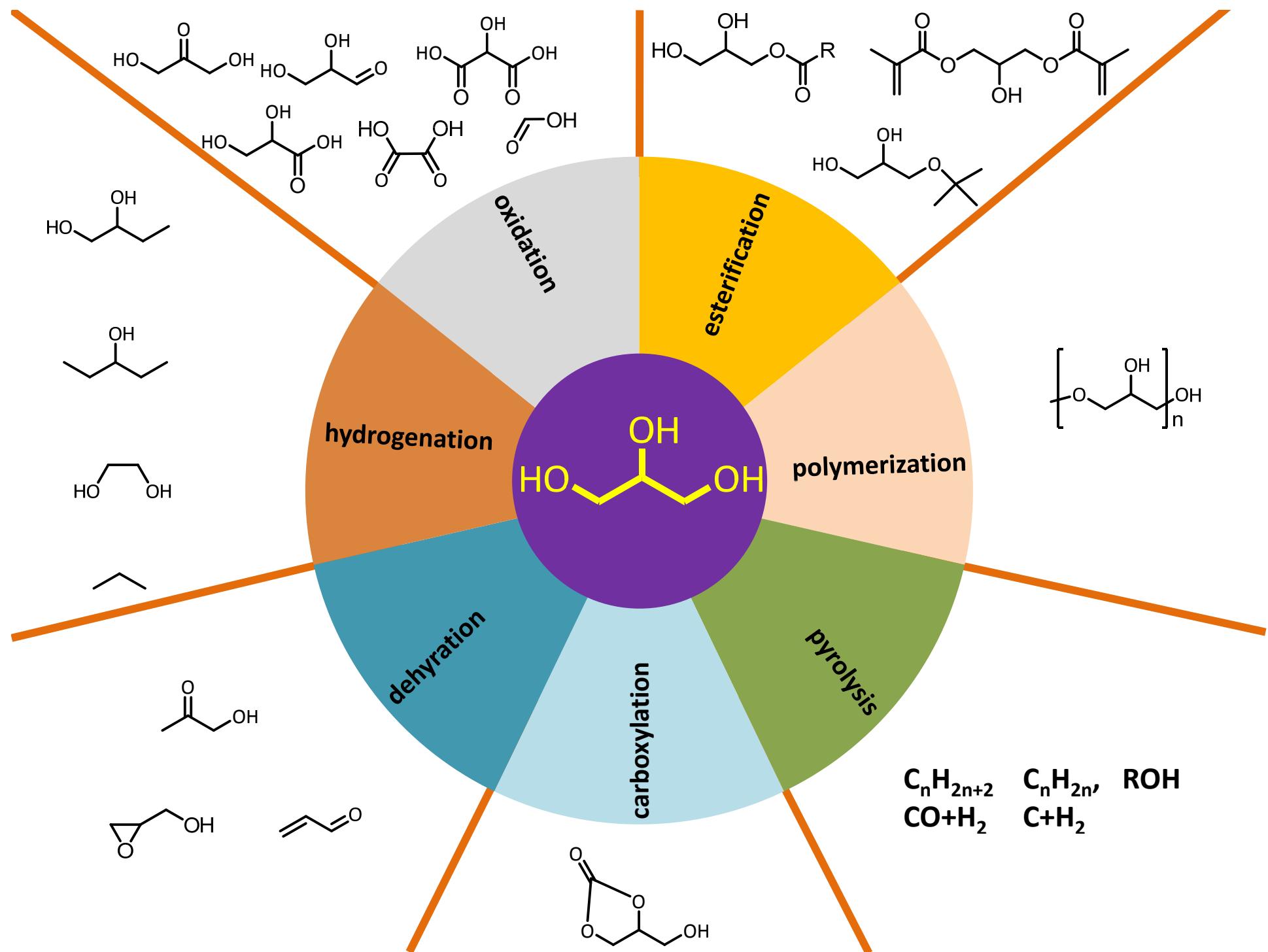


# 催化转化甘油已成为研究热点



数据来自 SciFinder，关键词：glycerol and catalysis

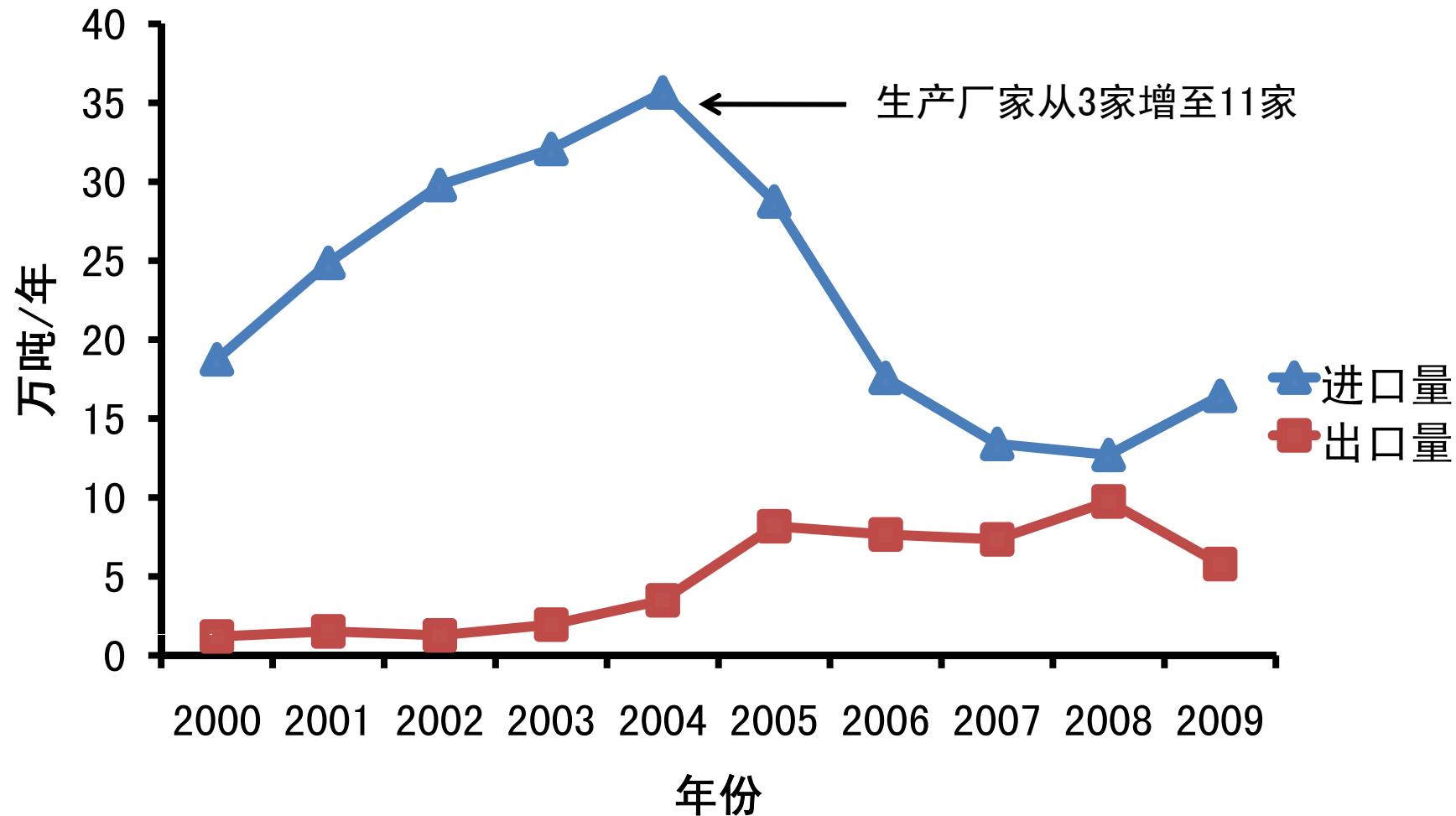
# 丙三醇制备化学品



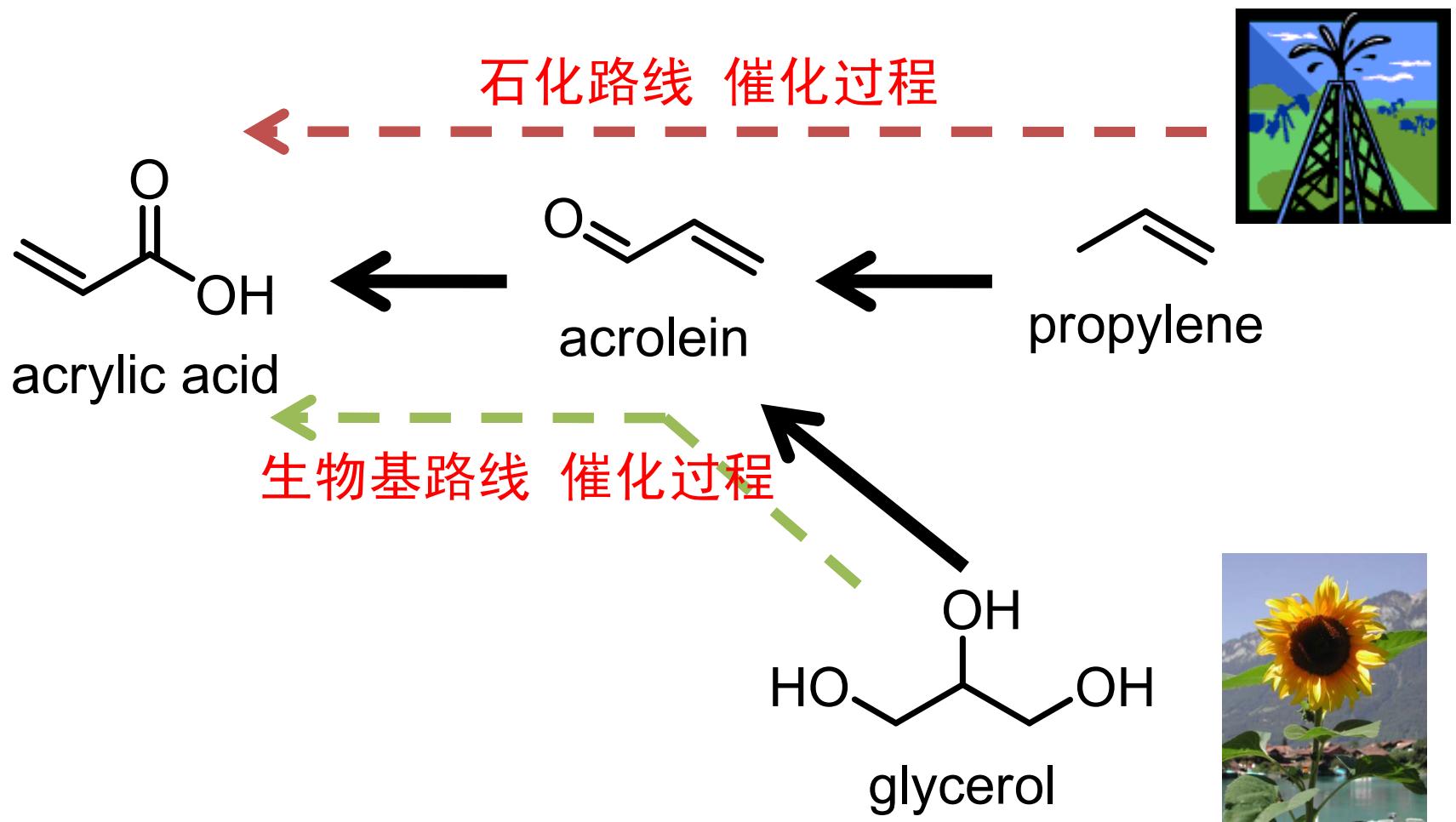
# 国内丙烯酸年总产约250万吨

企业名称	地点	丙烯酸 (万吨/年)	丙烯酸酯 (万吨/年)
北京东方	北京市	8. 0	8. 0
浙江卫星	浙江嘉兴	4. 0	4. 5
沈阳蜡化	沈阳市	8. 0	12. 0
江苏裕廊	江苏盐城	20. 5	25. 0
吉林石化	吉林市	3. 3	3. 0
上海华谊	上海市	18. 6	22. 0
扬子-巴斯夫	南京市	16. 0	15. 5
台塑(宁波)	浙江宁波	16. 0	20. 0
开泰实业	山东淄博	4. 0	0. 6
正和集团	山东广饶	6. 0	6. 0
兰州石化	兰州市	8. 0	10. 0

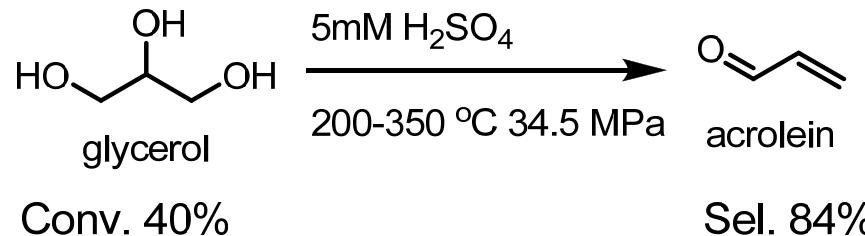
# 进口>出口



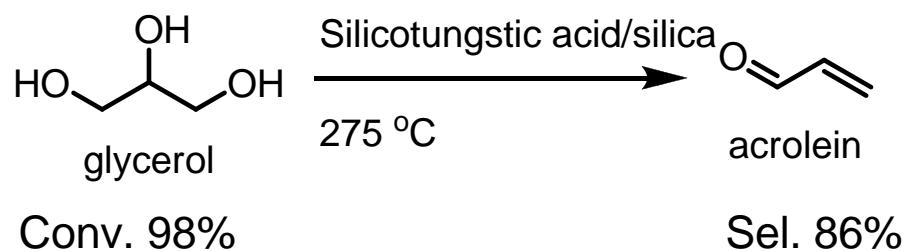
# 从可再生资源生产丙烯酸



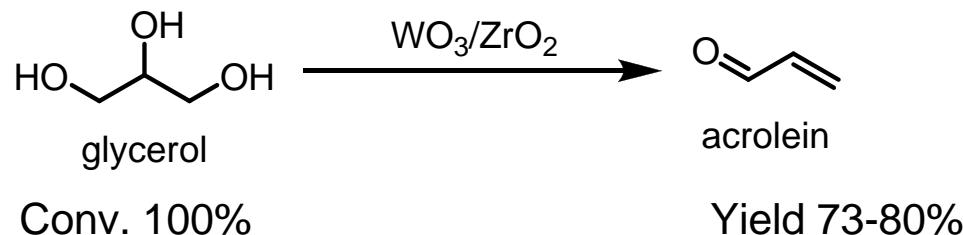
# 丙三醇脱水制备丙烯醛



S. Ramayya, A. Brittain, C. Dealmeida, W. Mok, M.J. Antal, *Fuel.* 66 (1987) 1364.

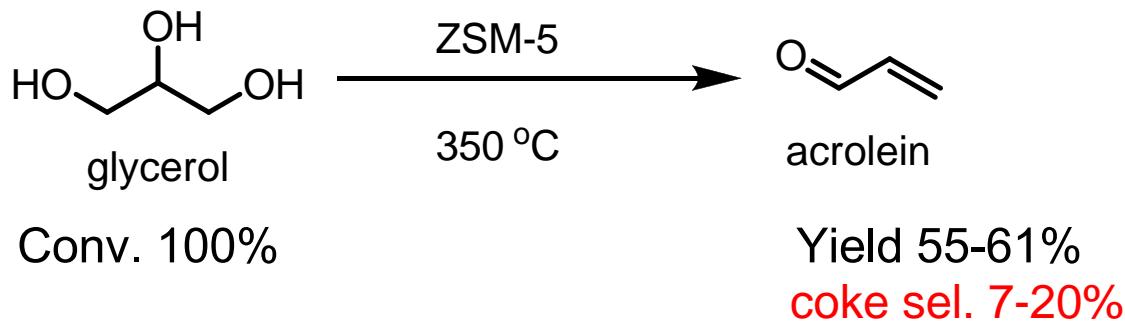


E. Tsukuda, S. Sato, R. Takahashi, T. Sodesawa, *Catal. Commun.* 8 (2007) 1349 .

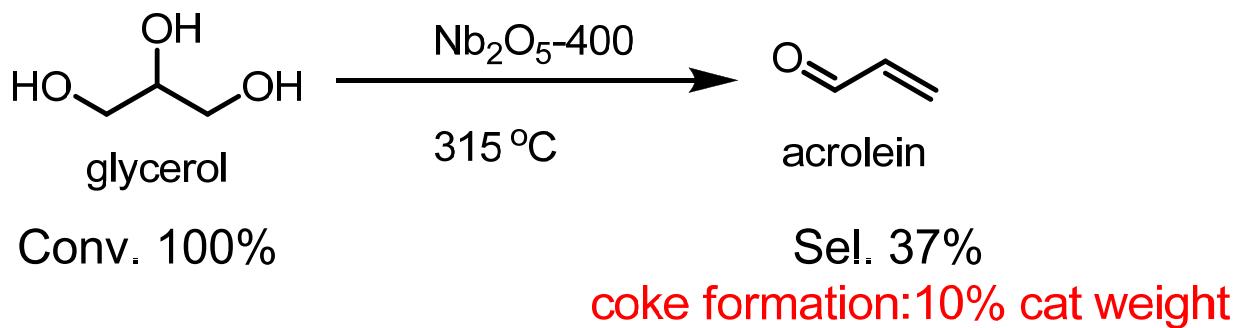


Dubois JL, et al. Arkema SA, WH (2005) F Patent 2,882,052; WH (2005) F Patent 2,882,053; WH (2005) F Patent 2,884,817; WH (2006) F Patent 2,884,818

# 脱水路线积碳严重

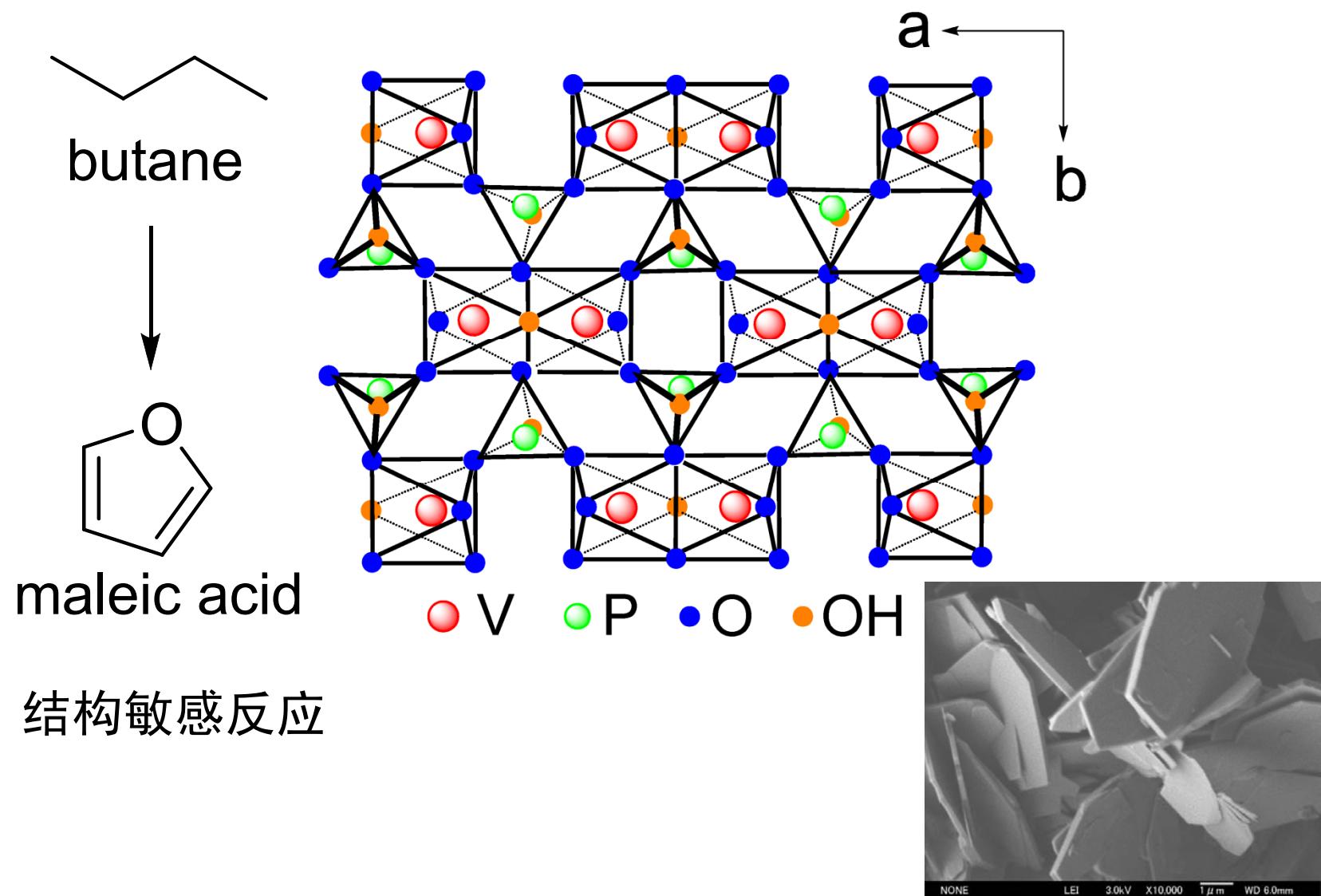


A. Corma, G.W. Huber, L. Sauvanauda, P. O'Connor, *J. Catal.* 257 (2008) 163

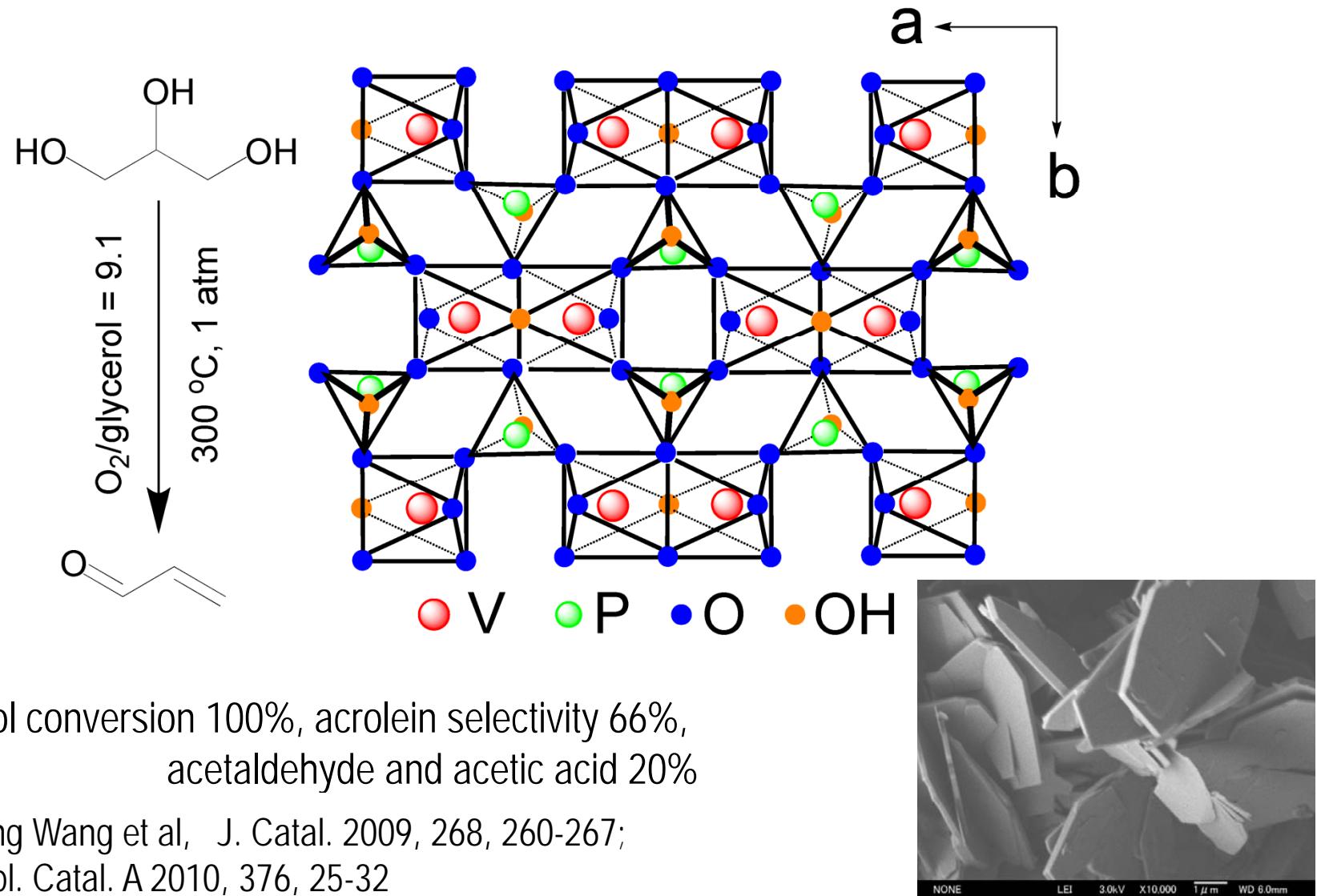


S.H. Chai, H.P. Wang, Y. Liang, B.Q. Xu, *J. Catal.* 250 (2007) 342

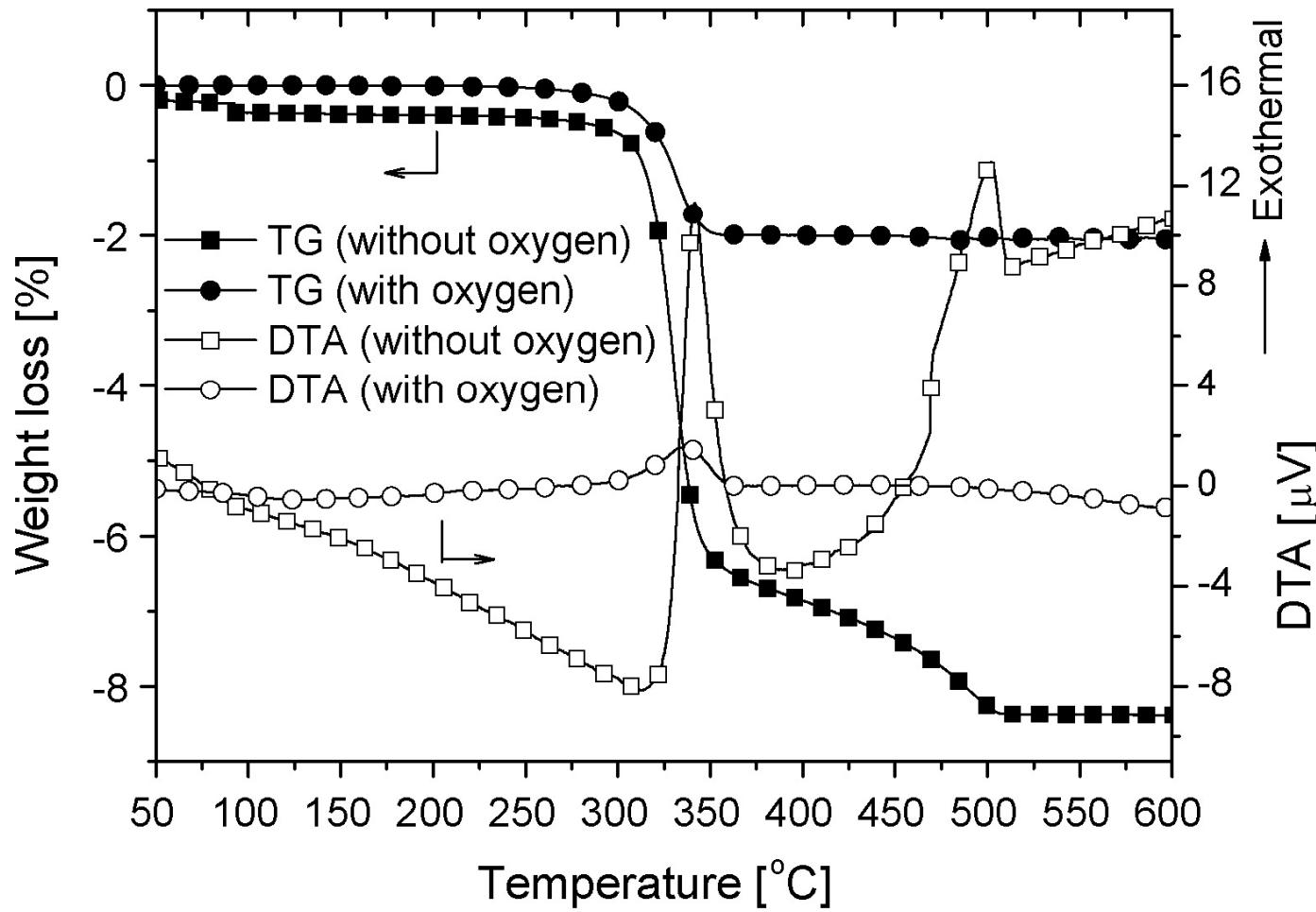
# V-P-O催化剂：丁烷制备马来酸酐



# V-P-O 催化剂：氧化脱水反应

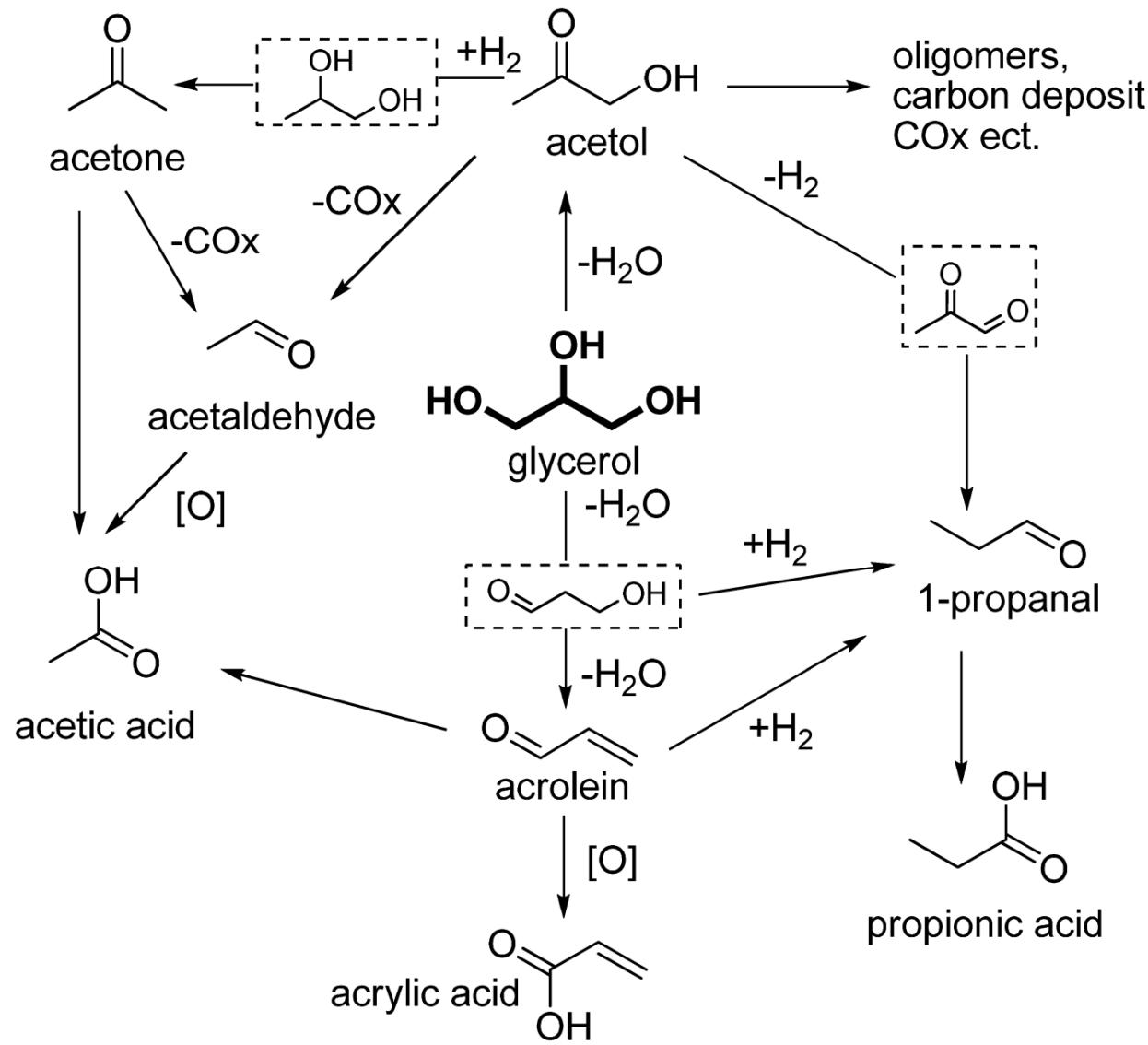


# 氧化气氛( $O_2/gly=9.1$ )减少积碳

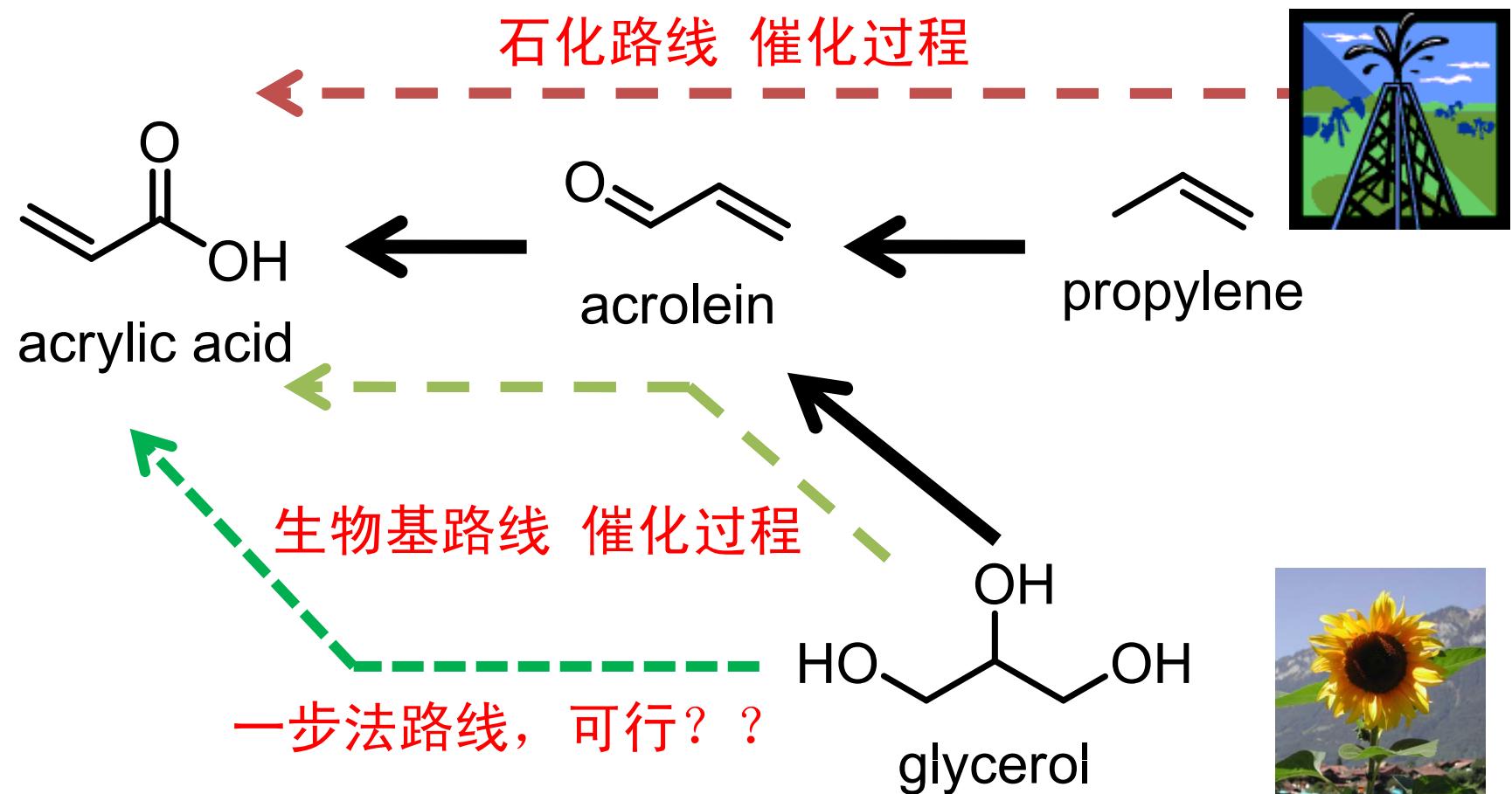


反应时间:10 h, 重时空速(WHSV): 2.5 h<sup>-1</sup>

# 复杂的反应路径



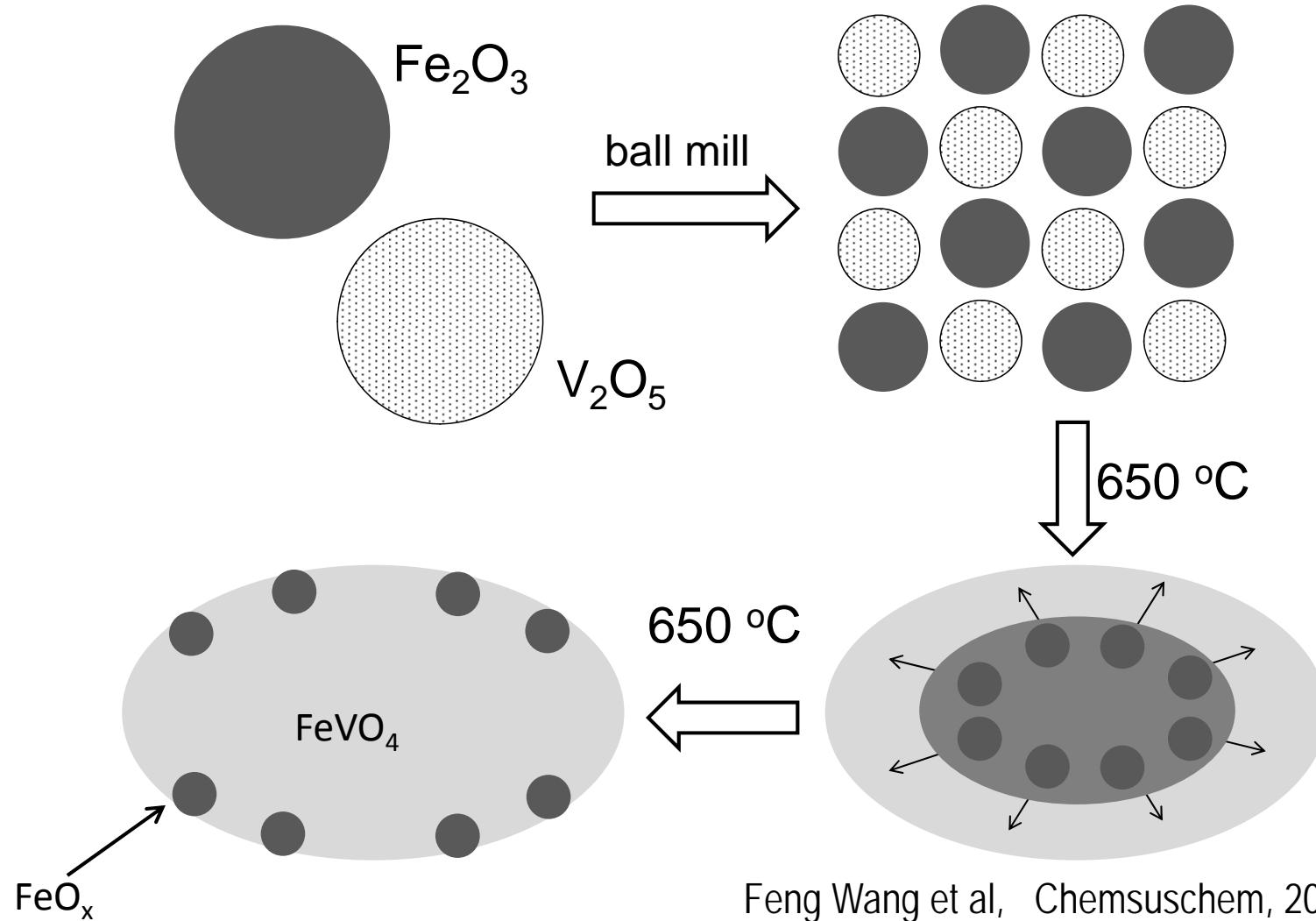
# 一步法从丙三醇制备丙烯酸



关键科学问题：具有氧化和脱水双功能的催化剂

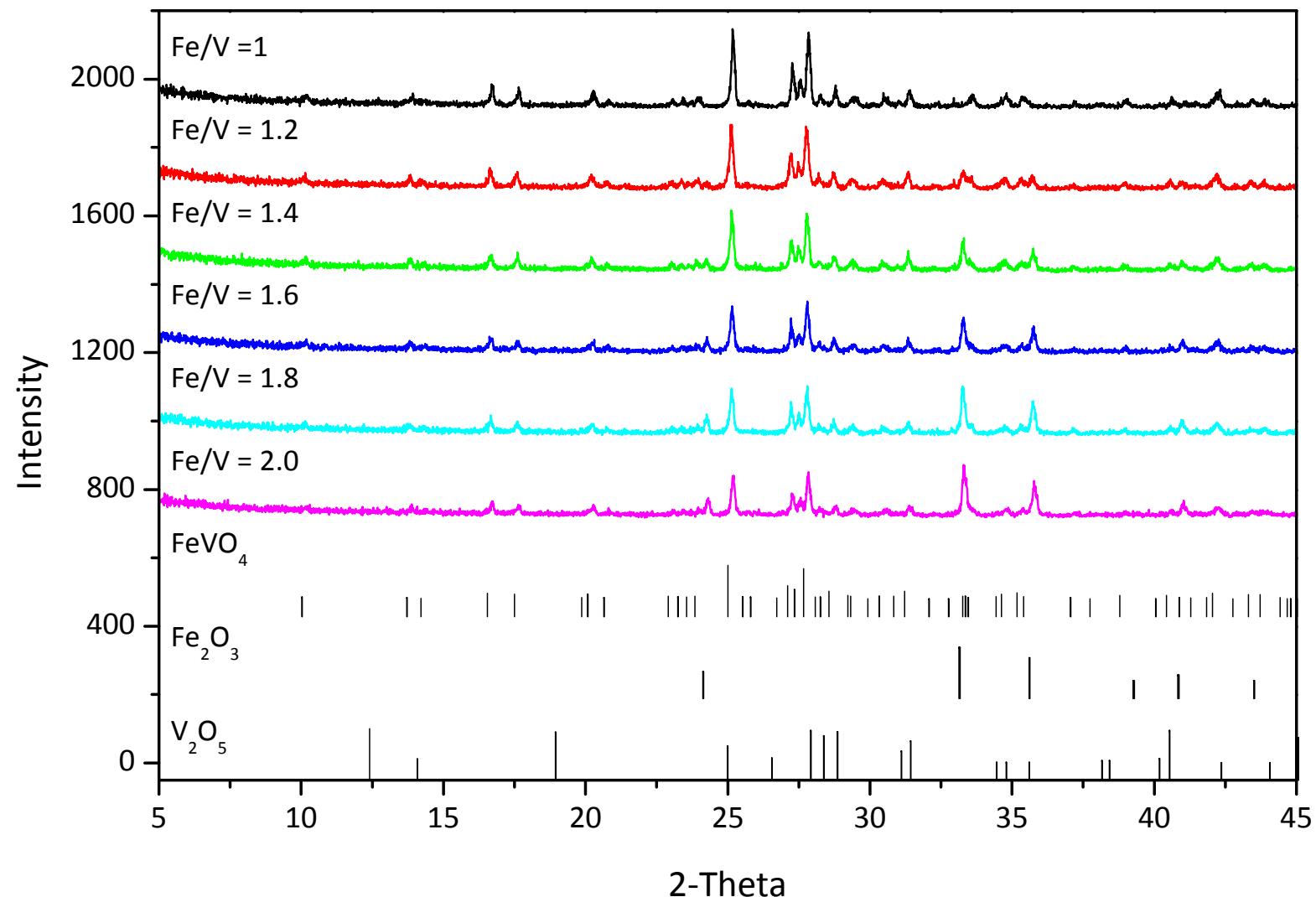


# 嵌入式催化剂 $\text{FeO}_x$ @ $\text{FeVO}_4$ 的制备

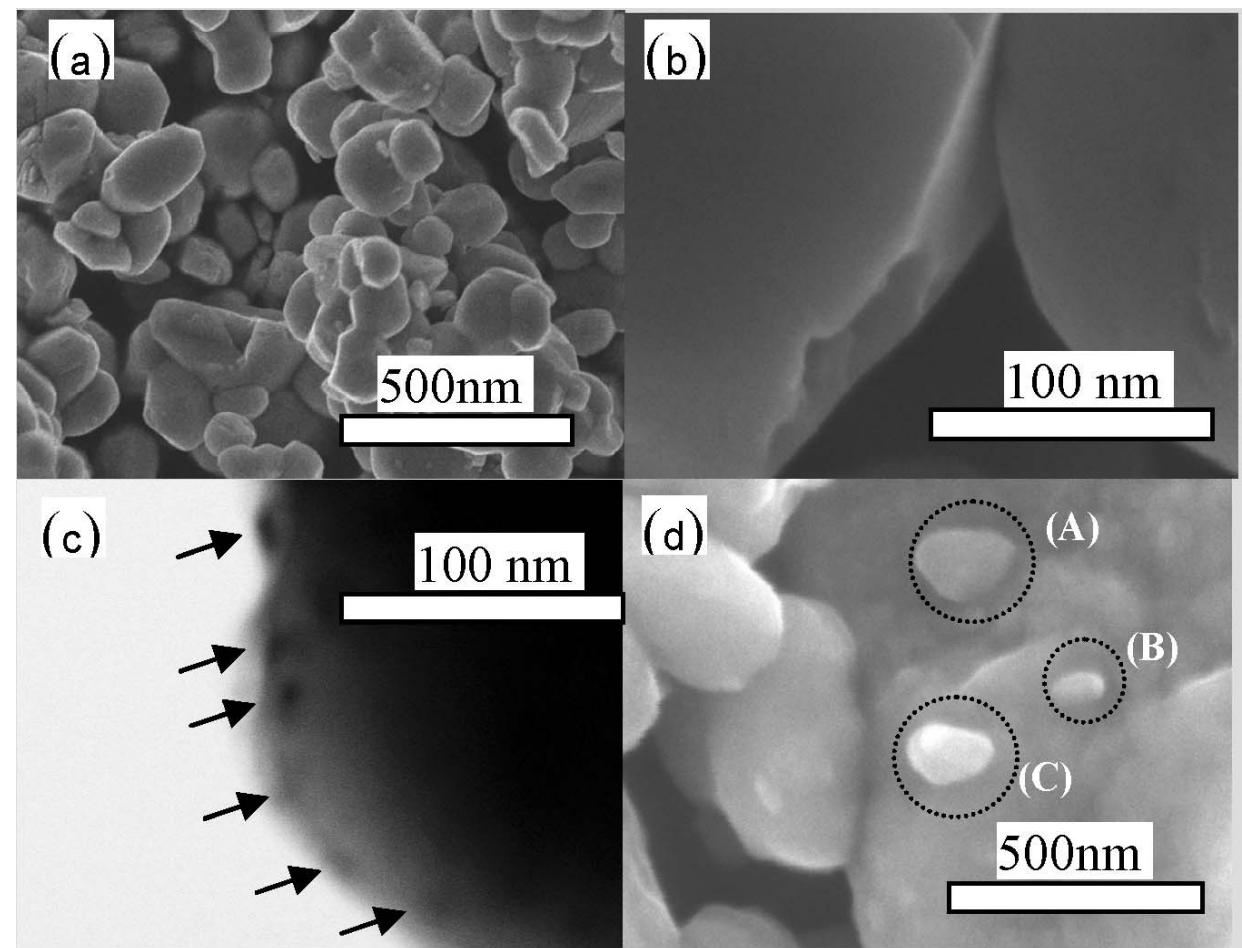


Feng Wang et al., Chemsuschem, 2010, in print

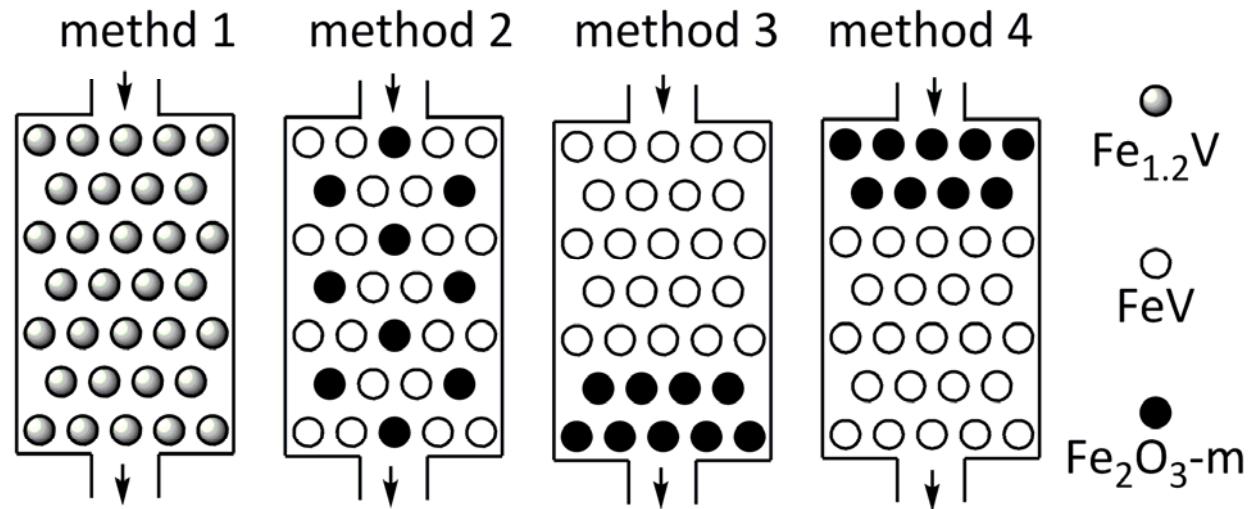
# 催化剂的结构与形貌



# 催化剂的结构与形貌

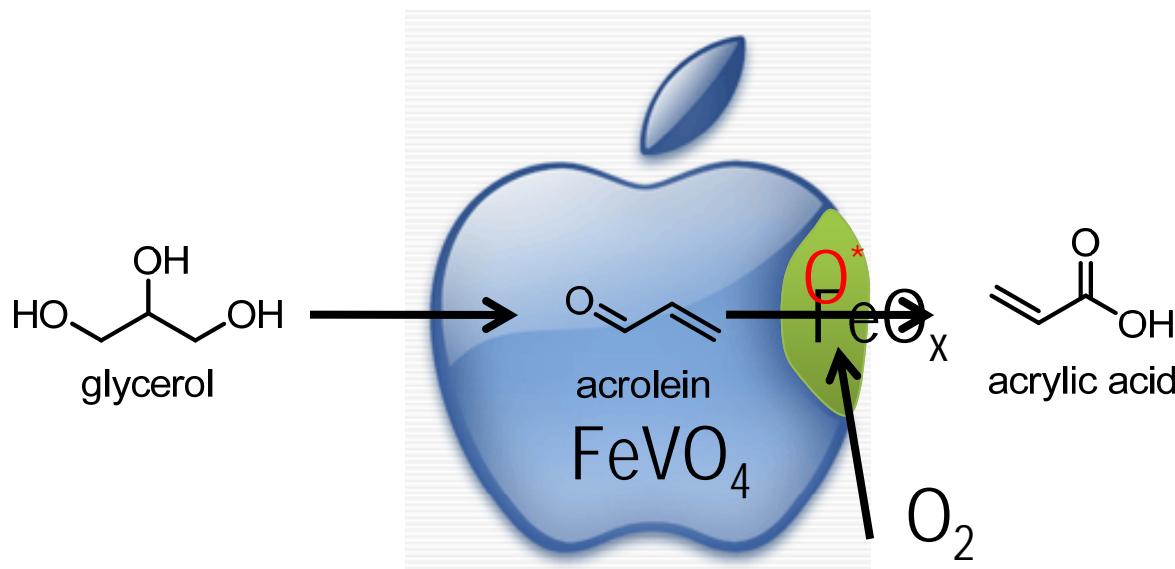


# 嵌入 $\text{FeO}_x$ 活性相催化性能优越



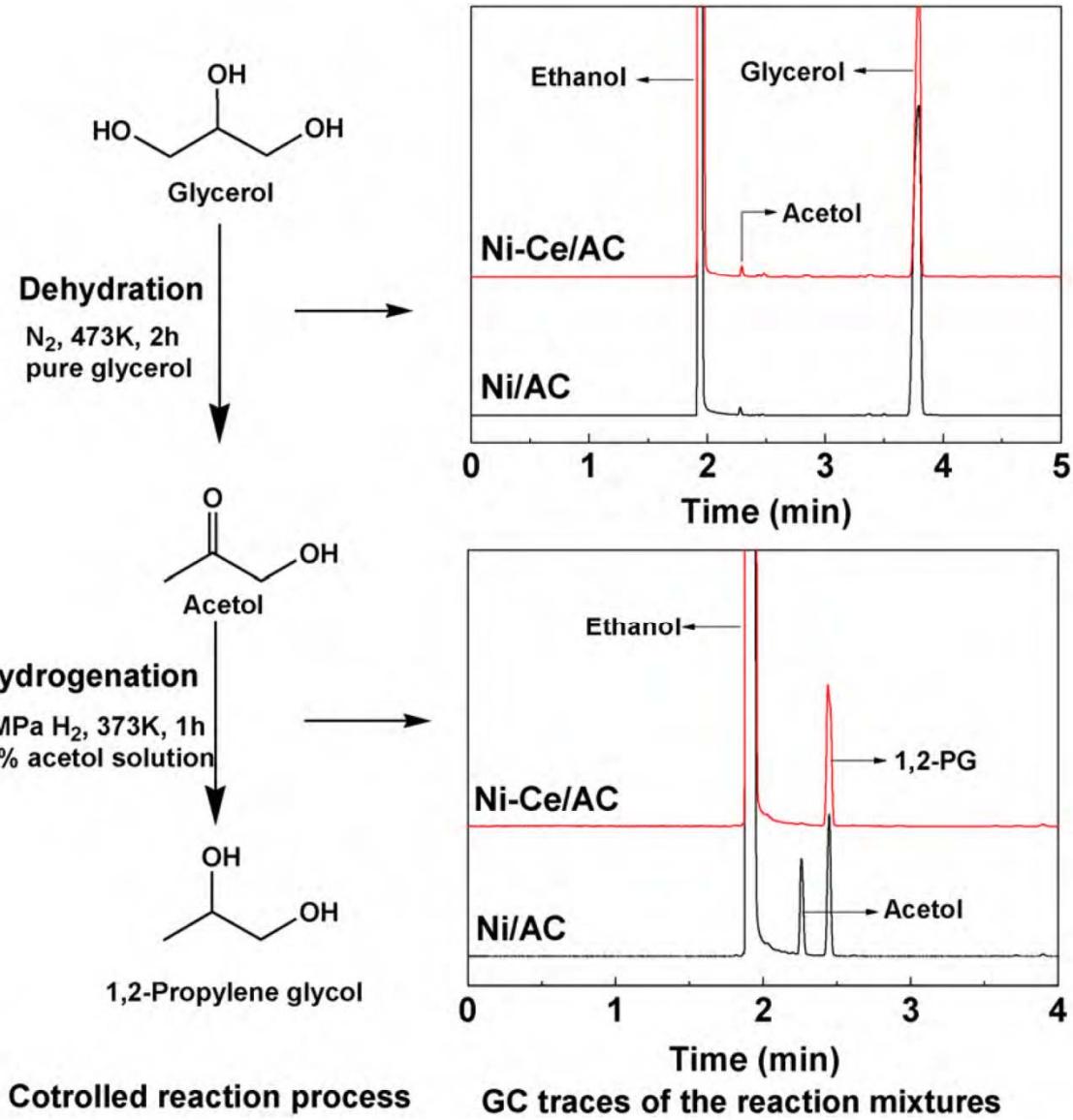
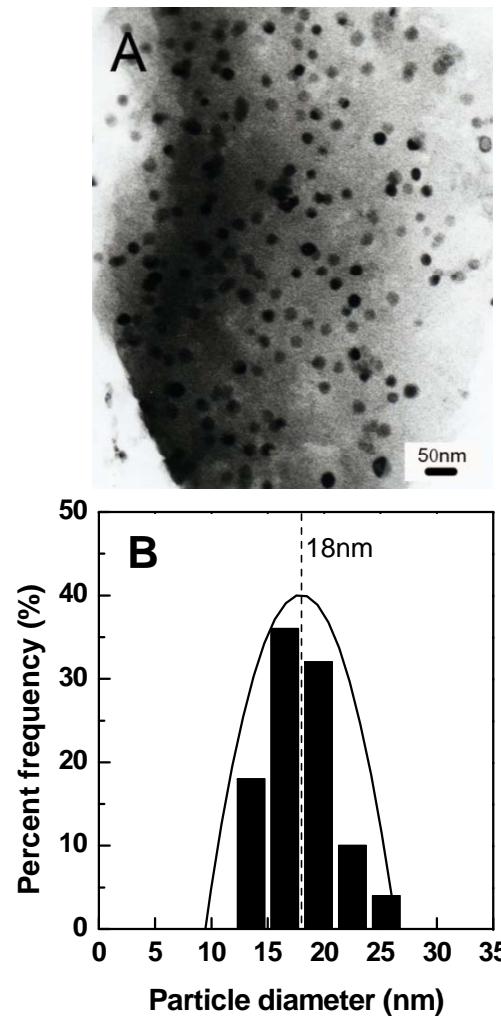
	Method1	Method2	Method3	Method4	负载 催化剂
甘油转化率	100	93	89	100	100
丙烯醛	28	30	50	40	5
丙烯酸	26	3	6	1	0
乙酸	24	21	10	13	64

# 可能的反应历程



双中心（脱水和氧化）嵌入式催化剂

# 催化加氢丙三醇制备1,2-丙二醇



# 催化加氢丙三醇制备1,2-丙二醇

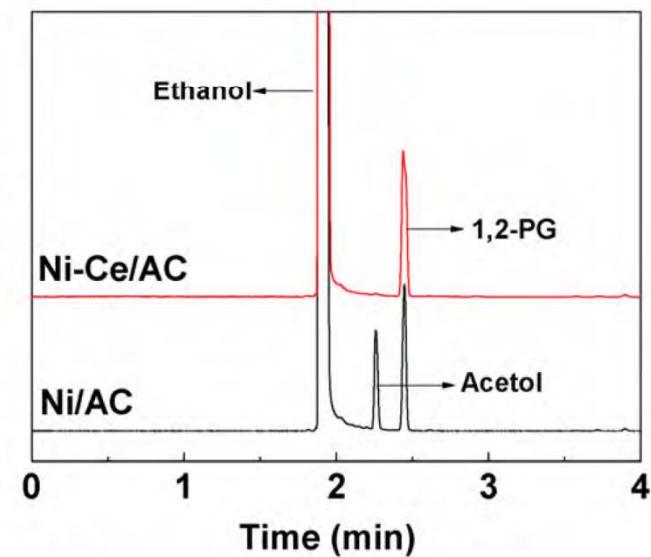
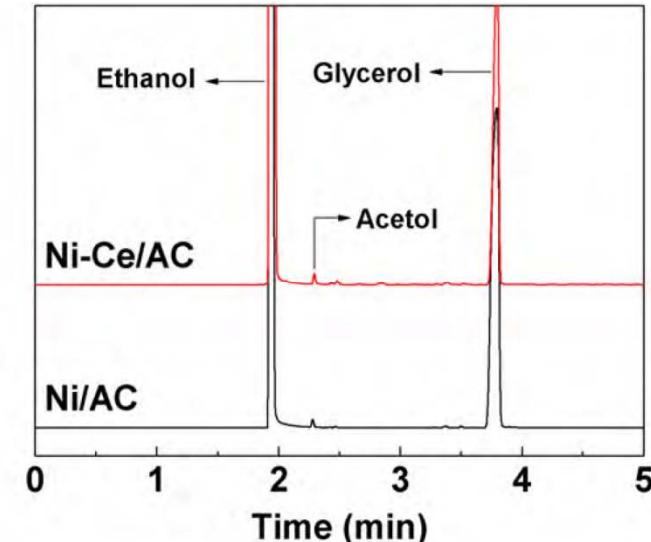
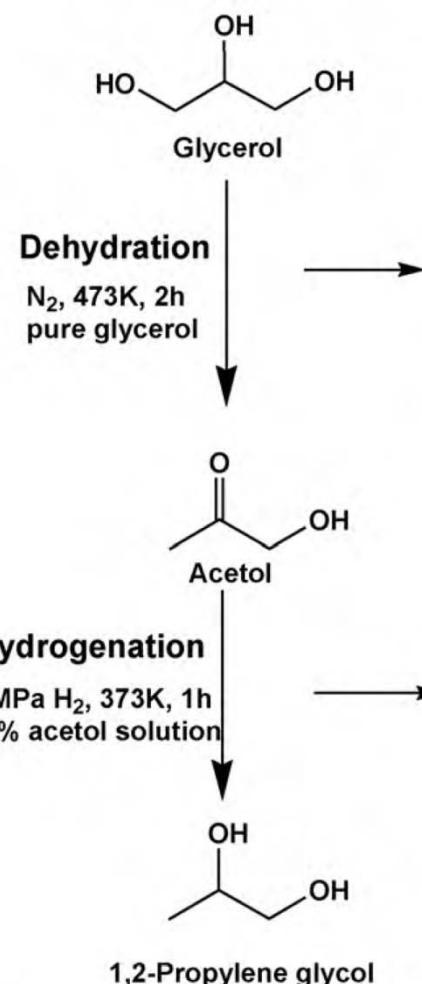
Supported nickel-Ce catalysts

Conversion of glycerol: 90.4%

Selectivity  
1,2-propylene glycol  
65.7%

Selectivity  
ethylene glycol  
10.7%

Weiqiang Yu, Jie Xu, et al  
Appl. Catal. A 383(2010) 73–78

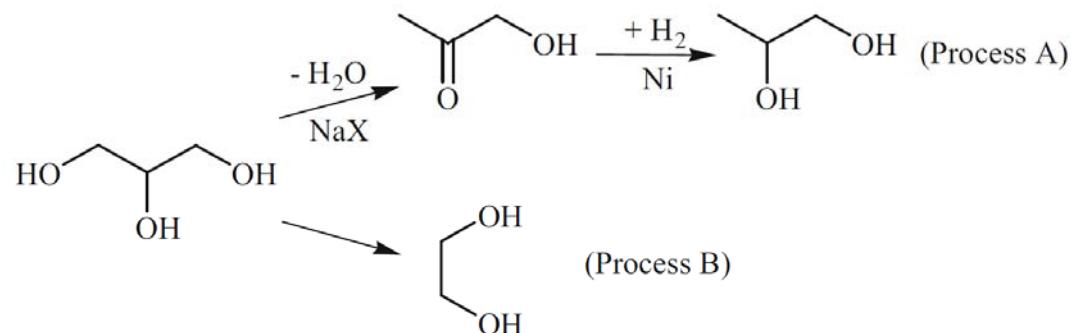


# 催化加氢丙三醇制备1,2-丙二醇

Entry	Catalyst	$S_{Ni}$ ( $m^2/g$ )	Conversion (%)	Selectivity (%)		
				1,2-PG	EG	Others <sup>a</sup>
1	Ni/NaMOR	21.2	14.0	56.7	13.3	30.0
2	Ni/NaZSM-5	9.8	47.8	9.4	13.0	77.6
3	Ni/NaA	10.4	65.3	46.8	14.1	39.1
4	Ni/NaX	25.5	94.5	72.1	11.1	16.8
5	Ni/SiO <sub>2</sub>	27.9	56.9	44.4	8.8	46.8
6	Ni/ $\gamma$ -Al <sub>2</sub> O <sub>3</sub>	22.2	97.1	44.2	7.5	48.3

Reaction conditions: 160 g 25 wt% glycerol aqueous solution, 2.0 g catalyst (metal-based), 200 °C, 6.0 MPa H<sub>2</sub>, 10 h, stirring speed: 500 rpm

<sup>a</sup> Others: ethanol, *n*-propanol, acetol, CO, CO<sub>2</sub> and CH<sub>4</sub>



# 结论

- 丙三醇氧化脱水反应，减少积碳；采用嵌入式催化剂，丙烯酸收率提高到26%
- 丙三醇加氢反应，镍基催化剂，伯碳加氢制备1,2-丙二醇收率68%

# 展望

- 丙三醇制备丙烯酸：双中心催化剂
- 丙三醇仲碳选择加氢制备1,3-丙二醇

谢谢大家！